

Concentrations and speciation of potentially toxic trace elements in waters of an urban estuary; Bayou Bienvenue, New Orleans, Louisiana

Karen Johannesson

*Department of Earth and Environmental Sciences, Tulane University,
New Orleans, Louisiana*

<http://tulane.edu/sse/eens/faculty/kjohanne/index.cfm>

kjohanne@tulane.edu

Acknowledgements

U.S. Geological Survey

**Long-term Estuary Assessment
Group (LEAG)**

**Tulane/Xavier Center for
Bioenvironmental Research**



Why Study Heavy Metals in Estuaries?

- **Estuaries serve as nurseries for marine organisms**
 - Commercially important shell- and fin-fishes
 - Shrimp, oysters, redfish
- **Estuaries can act as natural “filters” for heavy metals and other pollutants**



Bayou Bienvenue, Louisiana

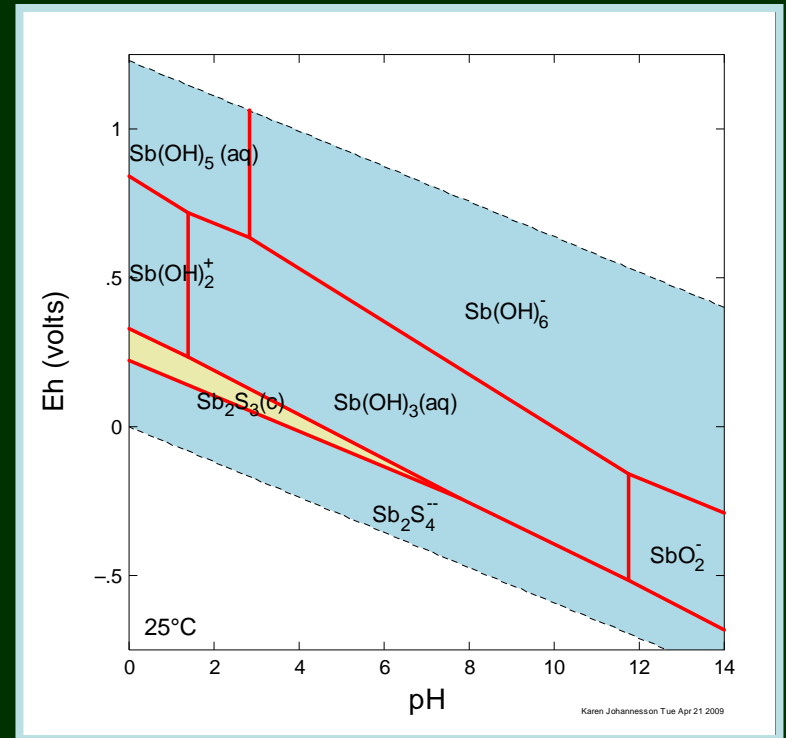
Why Study Heavy Metals in Estuaries?



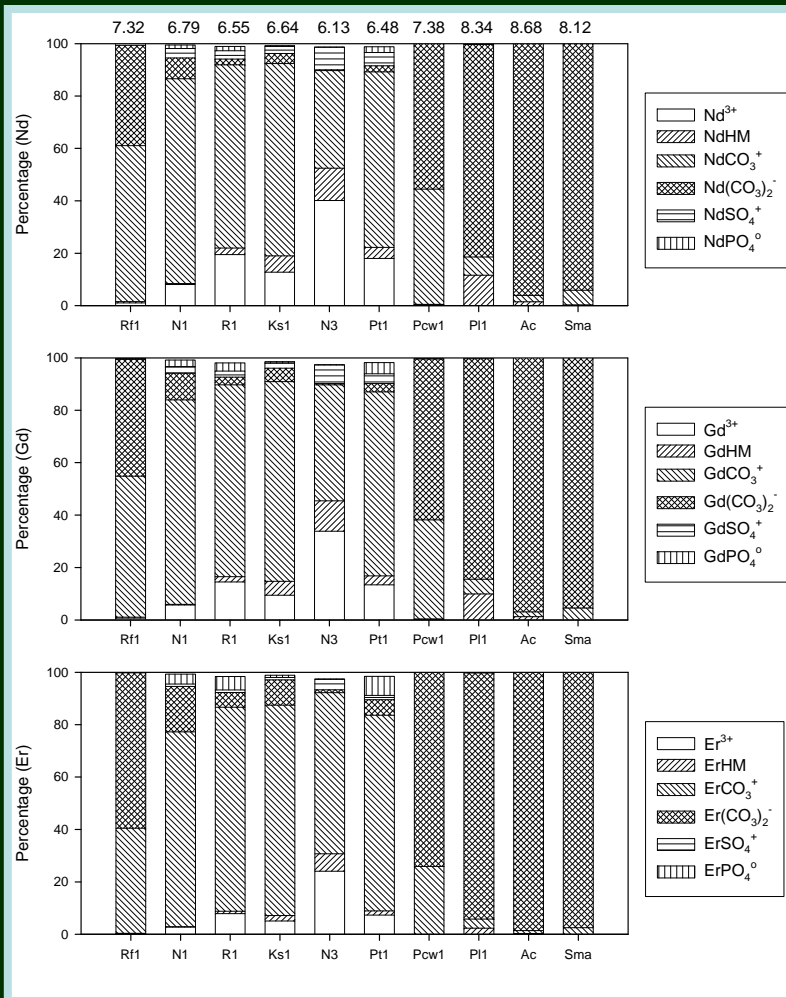
- Current paradigm → estuaries filter heavy metals
- Reality → little is actually known about the biogeochemical cycling and transport of heavy metals in Louisiana's numerous estuaries

Metal Speciation

- It's not enough to measure the concentrations of heavy metals
 - Tells us little about their bioavailability, toxicity, mobility
- Need to determine speciation



Metal Speciation



- The particular chemical form that an element exists in water

– Free ion, e.g., $[\text{Cu}^{2+}]_F$

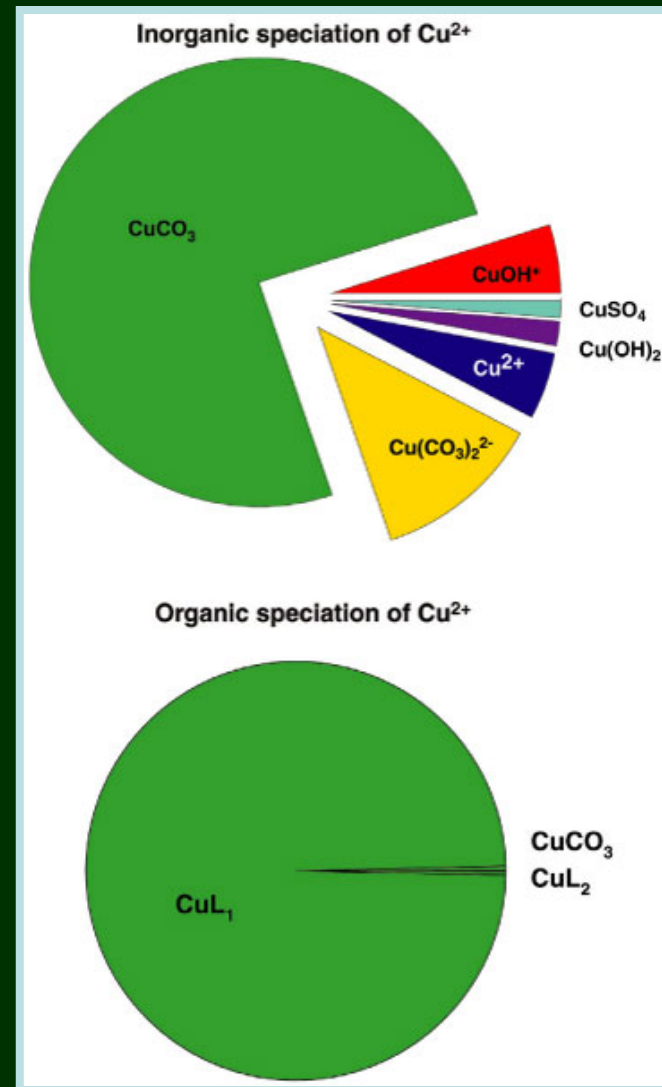
– Bound to organic ligands

– Complexed to inorganic ligands

– Different redox species, e.g., As^{3+} vs. As^{5+}

Metal Speciation

- Speciation controls trace element's:
- Bioavailability
 - Free-ion activity model
- Toxicity
 - As^{3+} is 10 – 60 more toxic than As^{5+}
- Effective solubility
- Mobilization and transport in the environment

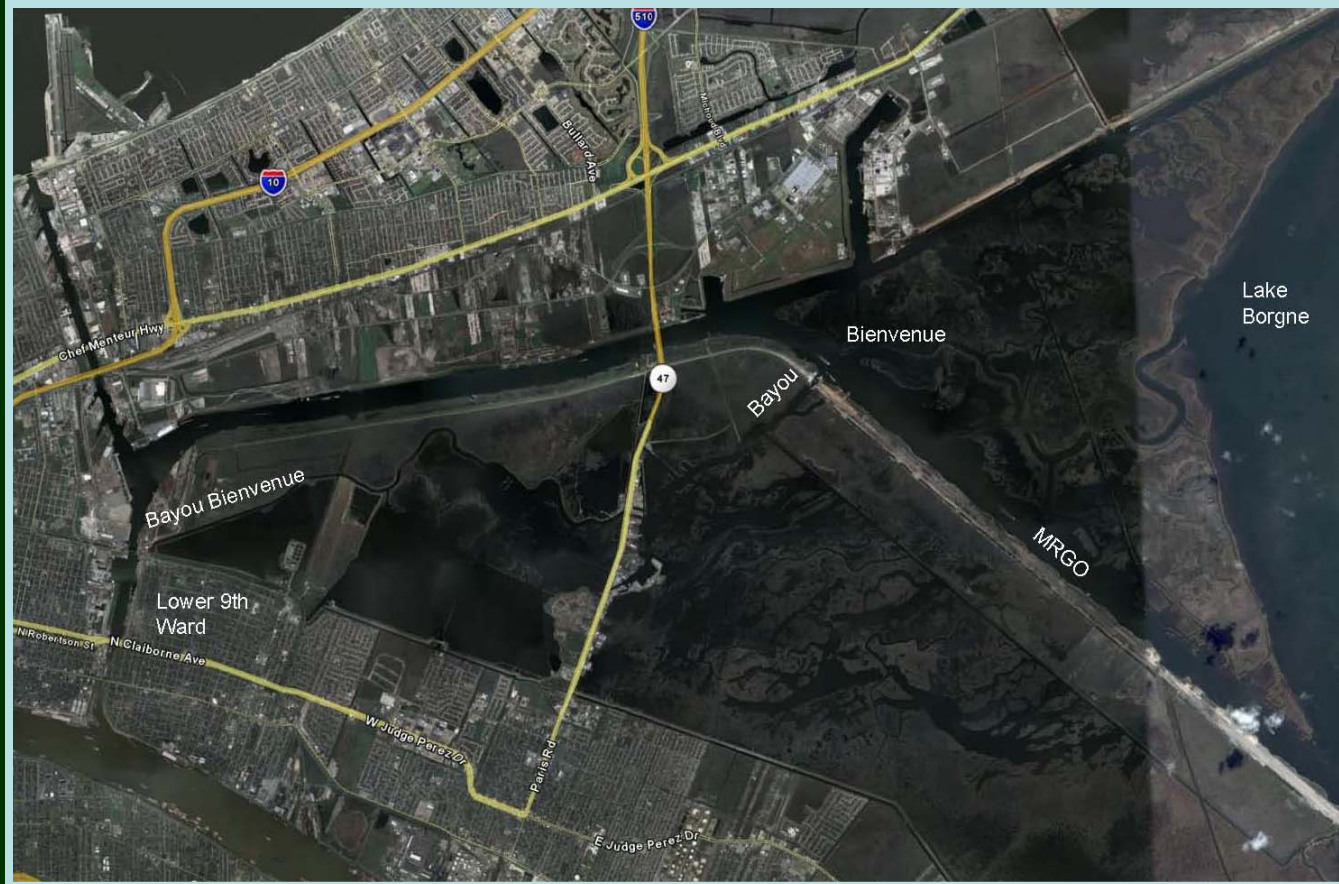


Methods/Description

- **Collect series of surface waters samples along Bayou Bienvenue**
- **Ultra clean trace element techniques**
- **Clean hands – dirty hands**



Bayou Bienvenue

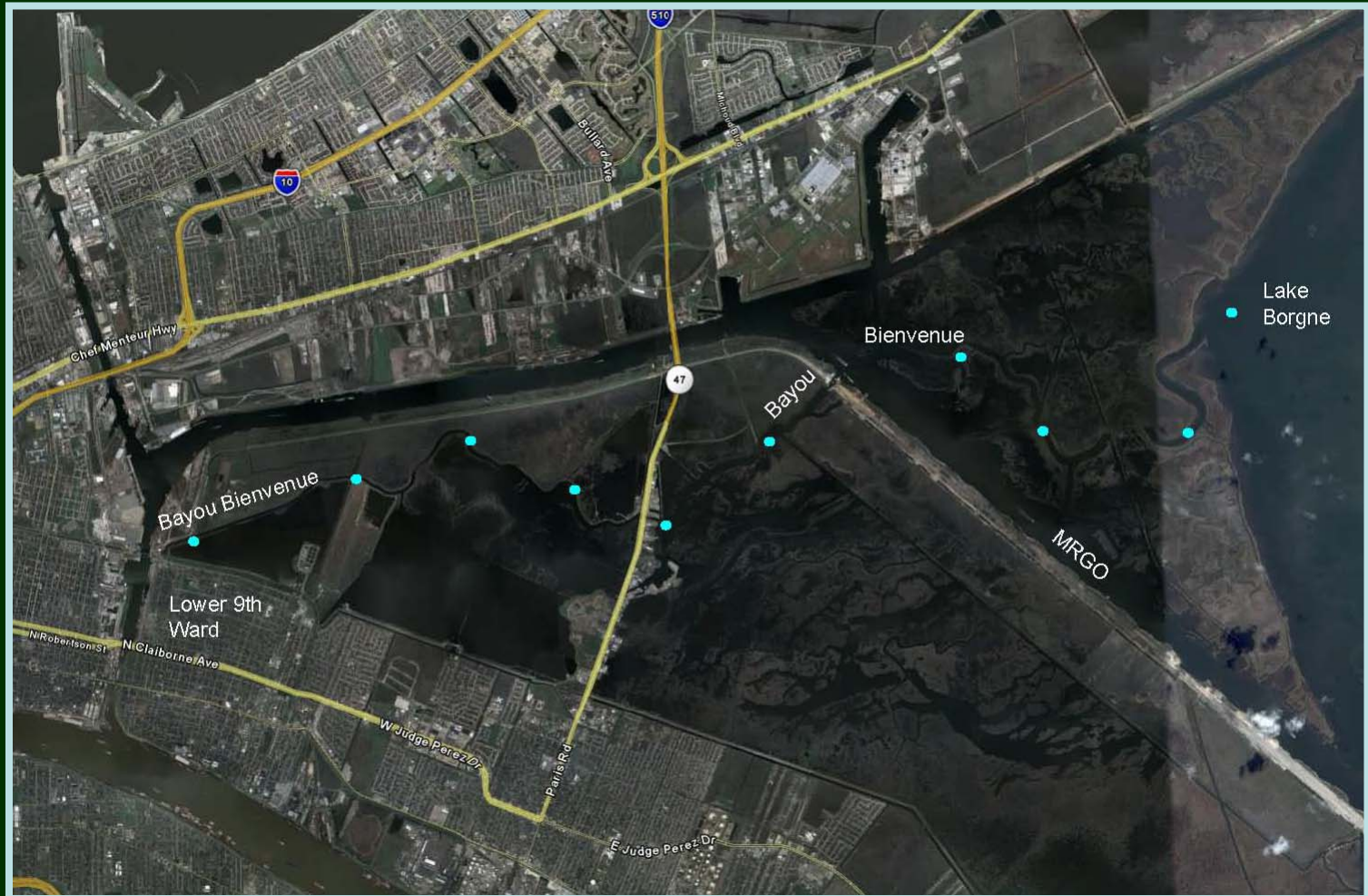


Bayou Bienvenue



Dr. Alex Kolker from LUMCON will assist us with boat time

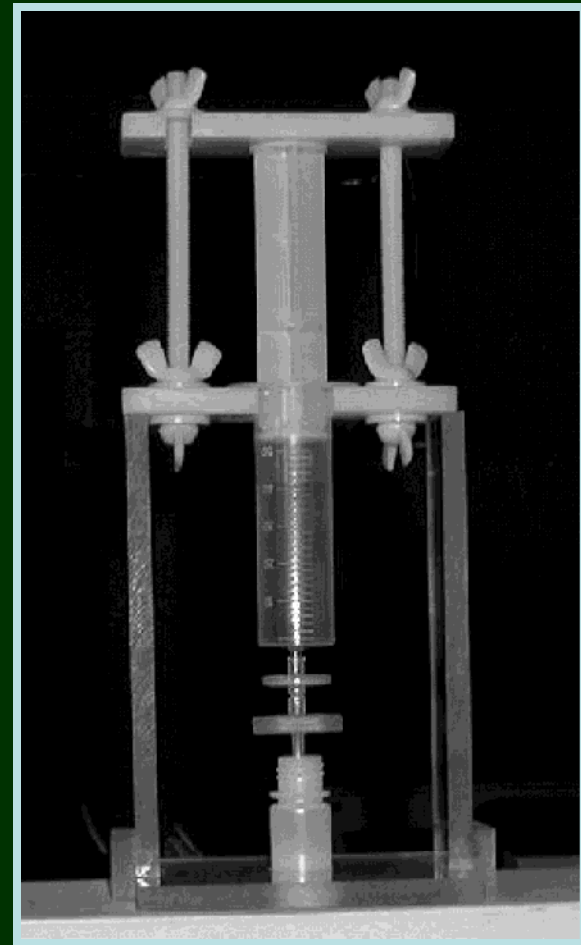
Bayou Bienvenue



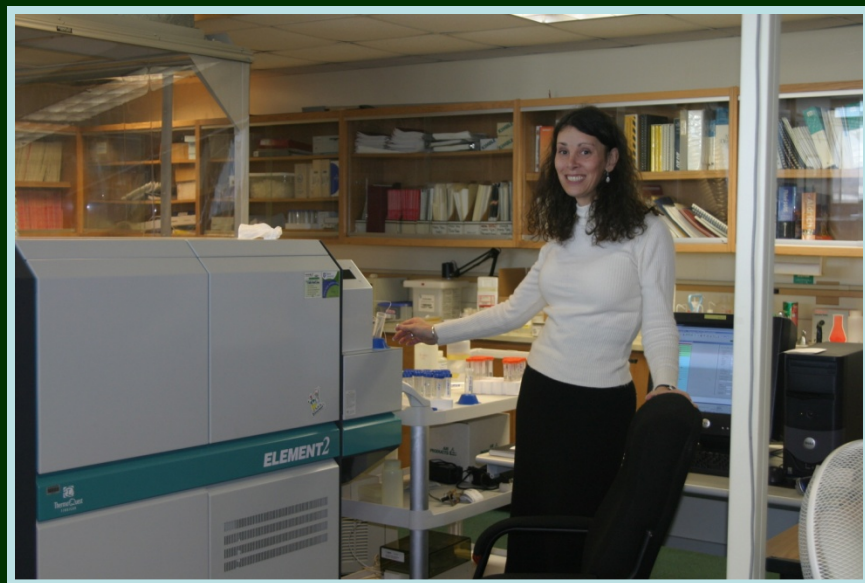
Blue dots show sampling sites on Bayou Bienvenue

Speciation Analysis

- **Filtration**
- **Unfiltered samples**
- **Filtered through 0.45 μm**
 - colloidal
- **Filtered through 0.02 μm**
 - “truly dissolved”



Trace Element Analysis



Magnetic Sector ICP-MS at Tulane University

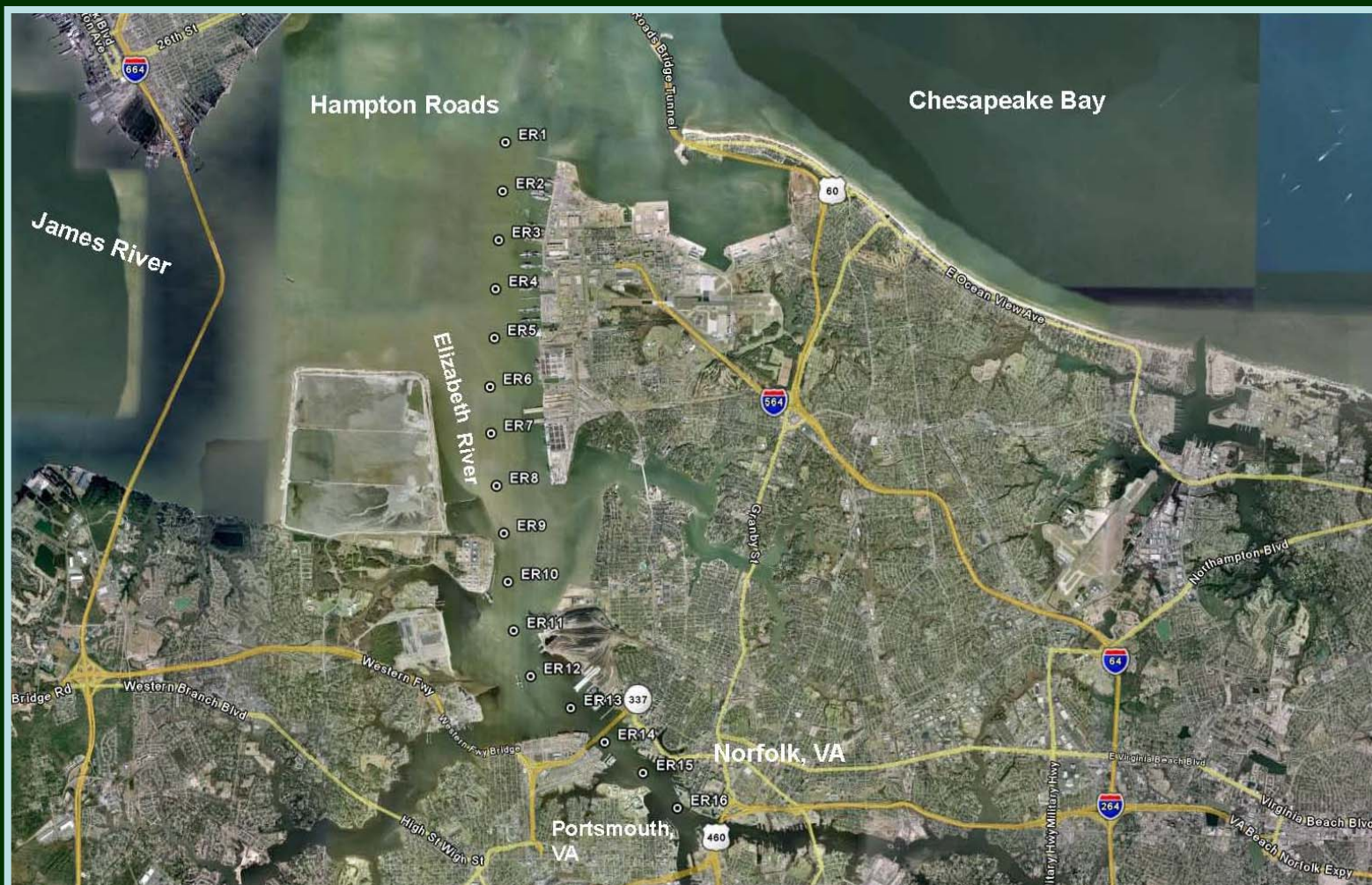
- Inductively Coupled Plasma Mass Spectrometry
- As, Se, Sb, Cr, Pb, Ni, Zn, Tl, Mo, W, V, U
- Fe, Mn
- Measure in each filtration aliquot

Organic carbon will be determined in each aliquot too.

Partition Coefficients

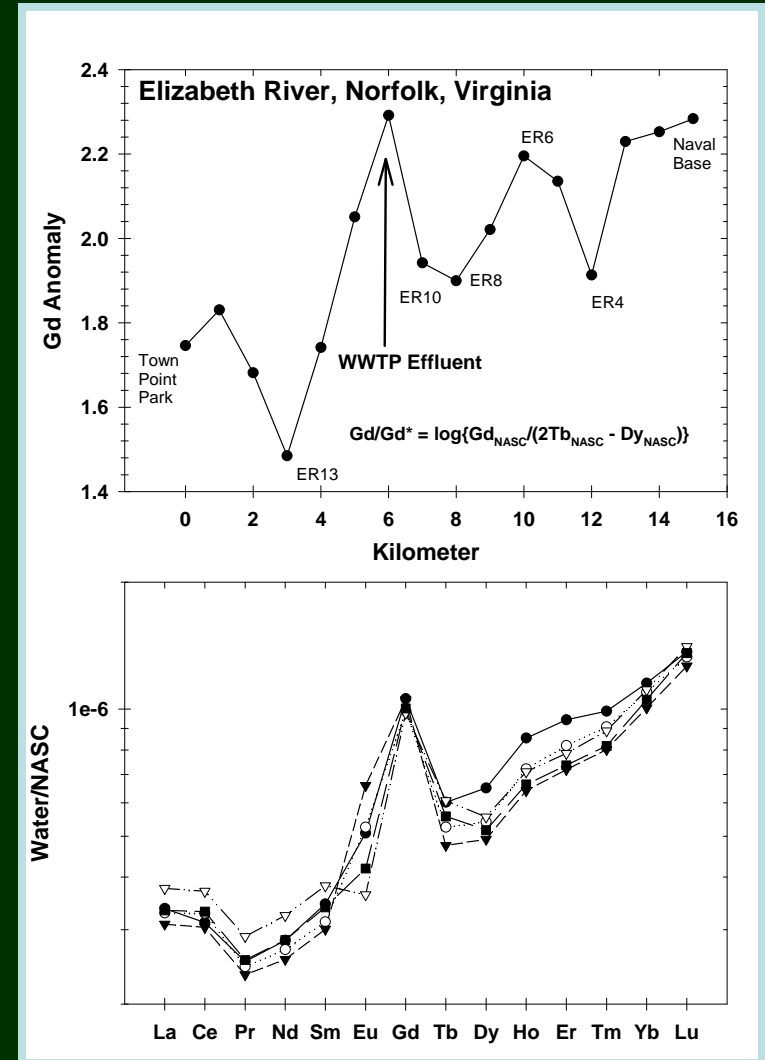
- $K_d^{\text{POC-DOC}} = \{[M_{\text{POC}}]/[\text{POC}]\}/[M_{\text{DOC}}]/[\text{DOC}]$
- $K_d^{\text{COC-DOC}} = \{[M_{\text{COC}}]/[\text{COC}]\}/[M_{\text{DOC}}]/[\text{DOC}]$
- [POC], [COC], & [DOC] are the concentration of particulate, colloidal, and dissolved organic carbon, respectively.
- $[M_{\text{POC}}]$, $[M_{\text{COC}}]$, & $[M_{\text{DOC}}]$ are the concentration of individual trace elements associated with different size fractions of OC.
- $M_{\text{POC}} > 0.45 \text{ } \mu\text{m}$
- $0.45 \text{ } \mu\text{m} \geq M_{\text{COC}} \geq 0.02 \text{ } \mu\text{m}$
- $M_{\text{DOC}} < 0.02 \text{ } \mu\text{m}$

Lanthanide Series



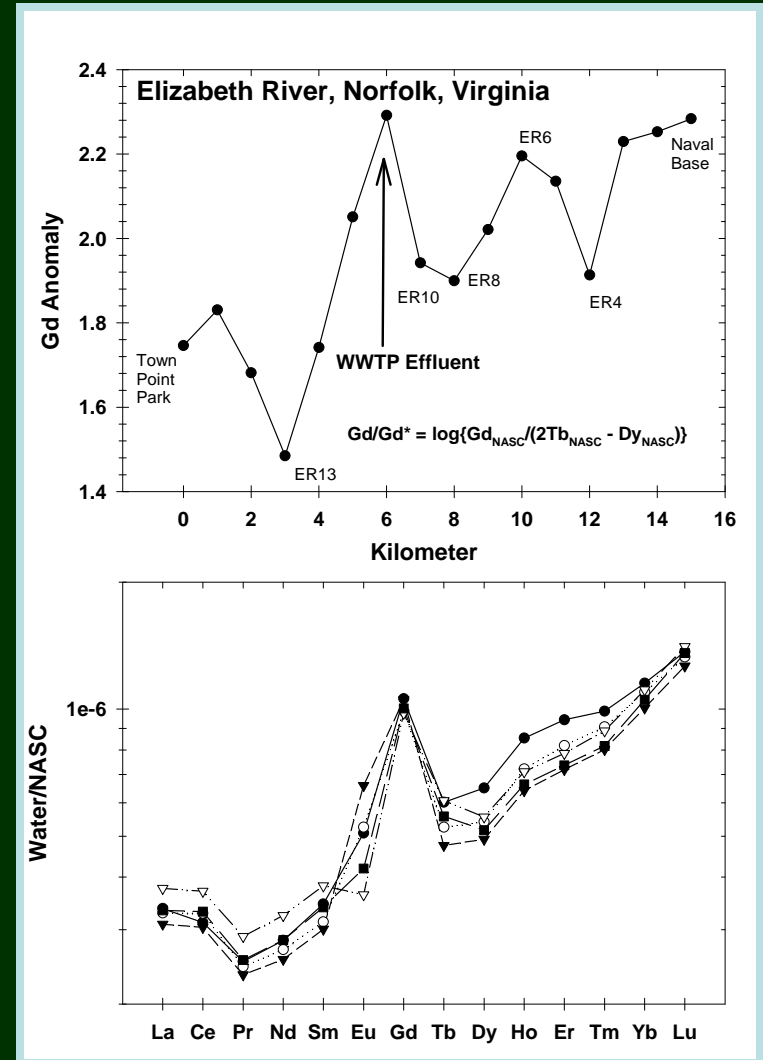
Anthropogenic Gd anomalies

- Gd is used in medical magnetic resonance imaging
- Gd has high magnetic moment
- Administered as:
 - Gadodiamide
 - gadopentetic acid
 - Gd-diethylenetriaminepentaacetate, i.e., Gd(DPTA)

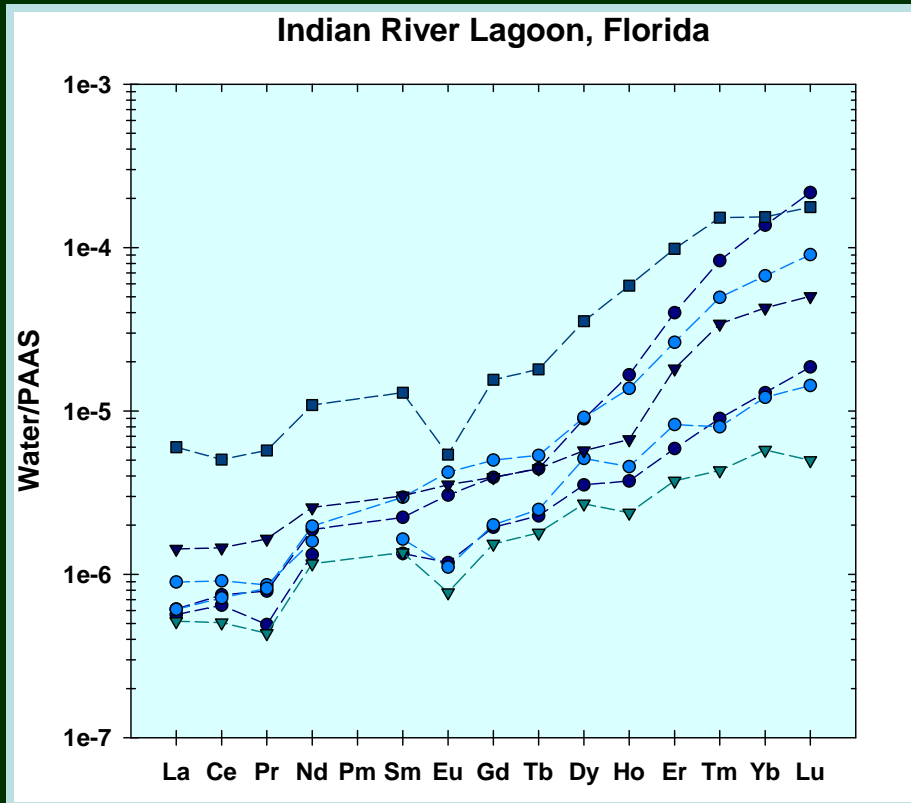


Anthropogenic Gd anomalies

- We will measure the lanthanide series elements in aliquots of our Bayou Bienvenue samples
- Direct measure of anthropogenic influences on these waters



Anthropogenic Gd anomalies



- Non-impacted estuaries do not have substantial Gd anomalies
- Indian River Lagoon in Florida

Johannesson et al. (in prep)